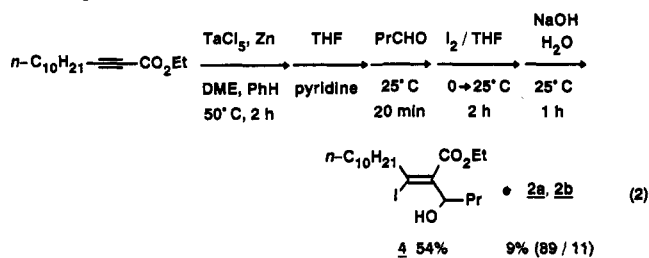


aldehydes was low. The observed reactivities and the β -selectivity could be attributed to the coordination of the nitrogen of amide to tantalum.¹³

Quenching the reaction mixture of a tantalum-ethyl tridecenoate complex and butanal with iodine in THF at 0 °C for 15 min and 25 °C for 2 h afforded β -iodo- α,β -unsaturated ester 4 in 54% yield¹⁴ along with untrapped allylic alcohols 2a and 2b in 9% combined yields (2a/2b = 89/11, eq 2). None of the α -iodo unsaturated ester was obtained. The carbon(sp²)-iodine bond of 4 is a clue to

develop further transformations.¹⁵



Supplementary Material Available: Experimental procedures and spectral data for all new compounds (13 pages). Ordering information is given on any current masthead page.

(13) When an acetylenic amide was added to the mixture of low-valent tantalum, the color of the mixture changed from greenish dark blue to ultramarine. Similar color change was observed after the addition of TMEDA to the low-valent tantalum.

(14) Reduction of β -iodo ester 4 with Et₃NH⁺HCO₂⁻ under palladium catalysis (Pd(PPh₃)₄) produced 2a in 87% yield. The following method was modified: Cacchi, S.; Ciattini, P. G.; Morera, E.; Ortari, G. *Tetrahedron Lett.* 1986, 27, 5541.

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Enantiocontrolled Total Syntheses of (-)-Physovenine and (-)-Physostigmine

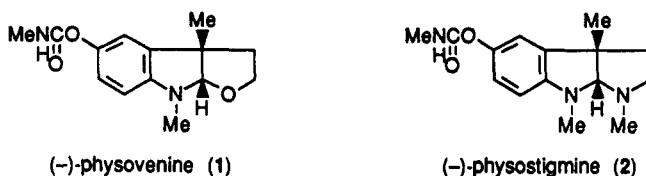
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Summary: Enantiocontrolled total syntheses of the Calabar bean alkaloid (-)-physovenine (1) and (-)-physostigmine (2) have been achieved in a concise manner starting from the optically active tricyclic enone 3 employing a Fischer indolization reaction under nonacidic conditions as the key step.

There is currently considerable interest in compounds having central stimulatory activity such as the anti-cholinergic Calabar bean alkaloids¹ due to their therapeutic potential in Alzheimer's disease¹ and cholinergic disorders.^{2,3} We report here the first total syntheses of (-)-physovenine (1) and a formal total synthesis of (-)-physostigmine (2), both members of the alkaloids of current interest, based on a new strategy exploiting a structurally biased polycyclic ketone 4 as a stereochemical control element in the key Fischer indolization step.



Alkylation of the optically active (-)-tricyclic enone⁴ 3, prepared from racemic dicyclopentadiene in a four-step

sequence of reactions including lipase-mediated resolution,⁵ afforded the monomethyl ketone 4 in 86% yield as a mixture of epimers. When this compound was refluxed with *p*-methoxyphenylhydrazine hydrochloride in aqueous pyridine⁶ (1:10), a facile diastereoselective reaction occurred to furnish the carbinol amine 7, mp 109–111 °C, [α]_D²⁵ -144.6° (c 1.95, CHCl₃), as a single product, in 82% yield. This compound is presumably generated via [3,3]-sigmatropic rearrangement of the diaza-1,5-diene intermediate 5 to afford the imine 6 via introduction of the aryl group from the convex face of the molecule. The imine 6 is then hydrolyzed under the reaction conditions to give the carbinolamine 7 instead of giving the pentacyclic indolenine.

On acetylation followed by methylation, 7 afforded the tertiary amide 9, [α]_D²⁵ -147.1° (c 1.21, CHCl₃), in 86% overall yield via 8, mp 163–164 °C, [α]_D²⁷ -100.3° (c 1.19, CHCl₃). Compound 9 was refluxed in *o*-dichlorobenzene to initiate a retro-Diels-Alder reaction to give the cyclopentenone 10, [α]_D³¹ -64.6° (c 1.41; CHCl₃), in 66% yield. The enone 10, on sequential one-flask ozonolysis, borohydride reduction, and periodate cleavage, furnished the lactol 14 in 62% yield via 11–13. Refluxing 14 in methanol containing a trace of hydrochloric acid caused concomitant deacetylation and cyclization to give the tricyclic amino acetal⁷ 15, [α]_D³² -96.2° (c 0.35, CHCl₃), in 71% yield. Treatment of 15 with boron tribromide^{3a} followed by carbamoylation of the resulting phenol 16 afforded (-)-physovenine⁷ (1), mp 126–127.5 °C, [α]_D^{30D} -90.0° (c 0.09,

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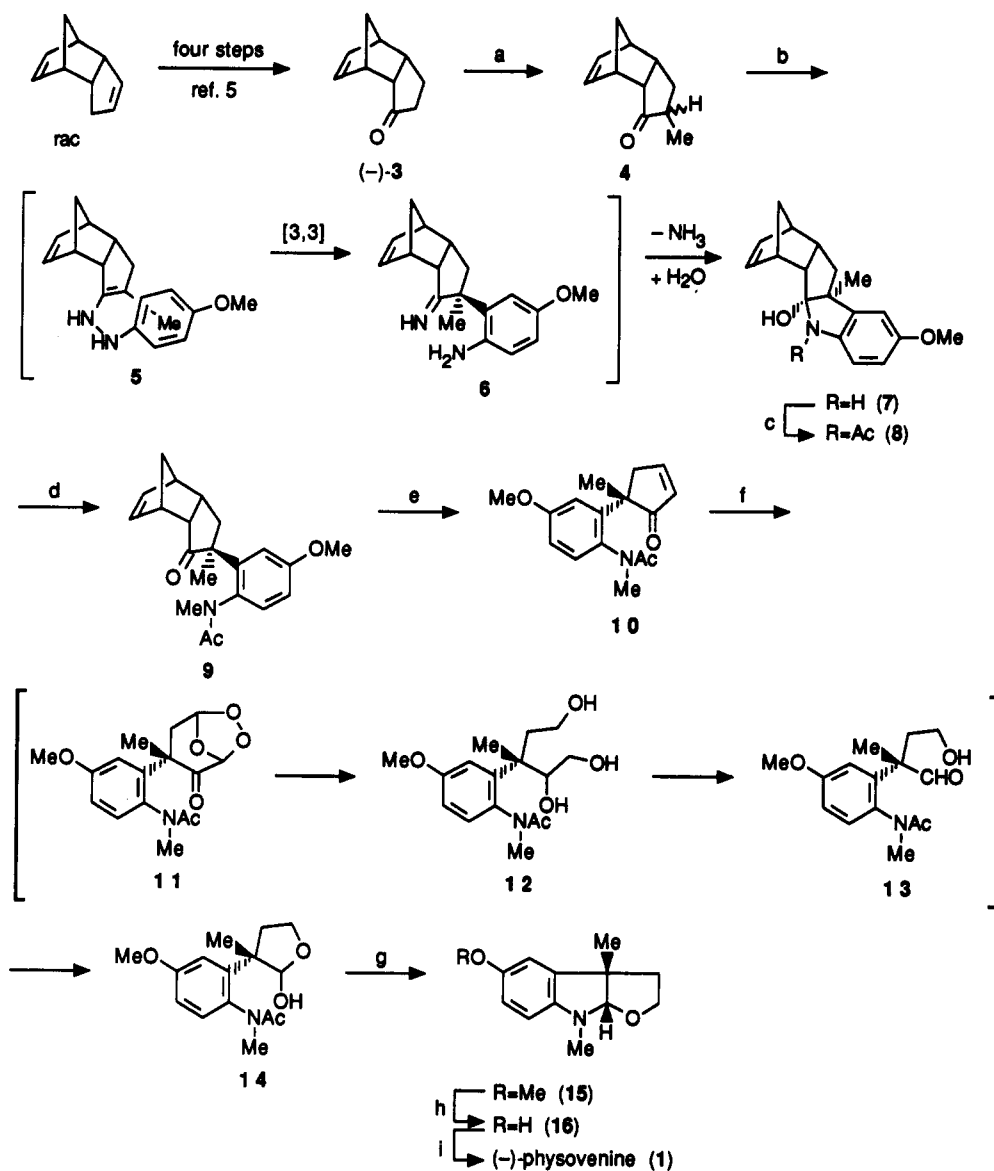
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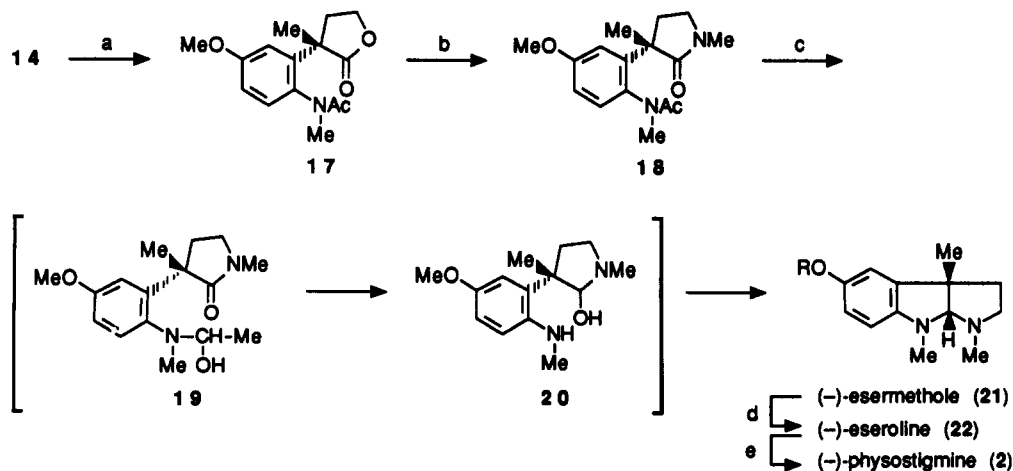
(6) Welch, W. M. *Synthesis* 1977, 645.

(7) Enantiomeric excess was determined to be $\geq 97\%$ by HPLC [CHIRACEL OJ, ¹PrOH-hexane (1:9)]. Spectral data (IR, ¹H NMR, and MS) of 15 were identical with those of racemic material: Shishido, K.; Shitara, E.; Komatsu, H.; Hiroya, K.; Fukumoto, K.; Kametani, T. *J. Org. Chem.* 1986, 51, 3007.

Scheme I^a



Scheme II^a



^aKey: (a) Ag₂CO₃ on Celite, benzene, reflux; (b) 40% aqueous MeNH₂, sealed tube 180 °C; (c) i, ¹Bu₂AlH, CH₂Cl₂, -78 °C then NH₄OH, ii, LAH, THF, reflux; (d) BBr₃, CH₂Cl₂, 0 °C → rt; (e) NaH (cat.), MeNCO, THF.

EtOH) [natural:⁸ mp 124-125 °C, $[\alpha]_{D}^{22.5} -92^{\circ}$ (EtOH)], in 84% overall yield. This route constitutes the first enantiocontrolled synthesis of the natural product.

On the other hand, oxidation of the lactol 14 by silver carbonate on Celite⁹ gave the lactone 17, $[\alpha]_{D}^{32} -69.9^{\circ}$ (c 0.30, CHCl₃), in 88% yield, which was transformed into lactam 18, $[\alpha]_{D}^{31} -72.4^{\circ}$ (c 1.47, CHCl₃), in 76% yield on heating at 180 °C with aqueous methylamine in a sealed tube. Upon exposure to diisobutylaluminum hydride at -78 °C followed by lithium aluminum hydride in refluxing THF, 18 furnished (-)-esermethole^{1,10} (21), $[\alpha]_{D}^{34} -134^{\circ}$

(c 0.41, benzene) [lit.^{3a} $[\alpha]_{D} -129^{\circ}$ (c 0.33, benzene)], directly in 34% yield, presumably via the carbinolamines 19 and 20. Since 21 has previously been transformed into natural (-)-physostigmine (2) in two steps^{3a} via (-)-eseroline (22),^{1,2} this sequence constitutes a formal synthesis of the natural product.

Acknowledgment. We are grateful to Professors Keiichiro Fukumoto and Kozo Shishido for a sample of physovenine.

Supplementary Material Available: Experimental details and spectroscopic data (IR and ¹H NMR) for compounds 1, 4, 7-10, 15, 17, 18, and 21 (6 pages). Ordering information is given on any current masthead page.

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(10) Enantiomeric excess was determined to be $\geq 98\%$ by HPLC [CHIRACEL OJ, ¹PrOH-hexane (1:9)]. Spectral data (IR, ¹H NMR, and MS) and TLC were identical with those of an authentic material.^{3a,b}

Articles

β -Lactams from Ester Enolates and Silylimines: Enantioselective Synthesis of the *trans*-Carbapenem Antibiotics (+)-PS-5 and (+)-PS-6

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A new synthetic route to the antibiotics (+)-PS-5 and (+)-PS-6 is described. The preparation involves a fully stereocontrolled reaction between the enantiomerically pure *N*-trimethylsilylimine of lactic or mandelic aldehyde and the lithium enolate of the *tert*-butyl butanoate or *tert*-butyl isovalerate, respectively. Conversion of the azetidiones obtained to 4-acetoxy derivatives via oxidative cleavage of the hydroxyethyl or hydroxybenzyl side chain and introduction of the necessary appendage in the position 4 of the azetidione ring, followed by assemblage of the bicyclic ring system, afforded the natural *trans*-carbapenems (+)-PS-5 and (+)-PS-6.

Introduction

The control of absolute stereochemistry is a central problem in the synthesis of biologically significant enantiomers of natural products. Of the large volume of literature on the synthesis of enantiomerically pure 3- and 4-disubstituted azetidion-2-ones as well as their corresponding bicyclic derivatives, the approach that leads to products of high enantiomeric purity either involves the use of an enantiomerically pure auxiliary, which is subsequently cleaved, or of an enantiomerically pure building block, which is retained in the target compound.¹

The cycloaddition reaction of ester enolates with aldimines has proved to be an effective method for preparing β -lactams.² Recently, in fact, we and others³ have demonstrated the synthetic usefulness of this reaction in the synthesis of thienamycin, using (*S*)-ethyl 3-hydroxybutanoate as the chiral nucleophilic component in the cycloaddition⁴ (Chart I). However, there are some β -lactam antibiotics bearing no stereogenic centres in the C-3 side chain of the azetidione ring. For instance, the carbapenems (+)-PS-5 and (+)-PS-6 have an ethyl and isopropyl group, respectively, in this position. In these case,⁵

in order to prepare enantiomerically pure compounds, the asymmetry can be incorporated in the electrophilic partner

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